

# Formation of $\text{Cr}_3\text{C}_2/\text{Ni-Cr}$ Alloy Layer by an Electron Beam Cladding Method and Evaluation of the Layer Properties<sup>†</sup>

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The trend toward more efficient utilization of energy and resource, as well as the need for advanced functionality, has led to a growing demand for materials with high erosion and corrosion resistance. Chemical coating methods using heavy metals are no longer a viable choice because of global environmental concerns and the threat of environmental pollution. Thus, there is a pressing need for new surface modification technologies that can form surface layers with superior functional properties. The authors have been developing an electron beam cladding method employing a high energy density electron beam and a powder feeder<sup>1-3)</sup>. In this report, a hard surfacing layers with high corrosion and erosion resistance were successfully formed on mild steel employing the electron beam cladding method with  $\text{Cr}_3\text{C}_2/\text{Ni-Cr}$  mixed alloy powder, and their properties were examined.

Figure 1 shows a schematics drawing of the experimental apparatus. A 30kW-class electron beam welder with acceleration voltage of 40kV was used as the heat source. The electron beam was focused by two magnetic focusing lenses to achieve a high energy density of over  $200\text{kW}/\text{cm}^2$  at a focal point when the output power was 1600W. A powder feeder designed to work stably under vacuum conditions supplied mixed powder. In order to form a hard surfacing layer with high corrosion and erosion resistance,  $\text{Cr}_3\text{C}_2/\text{Ni-Cr}$  alloy powder was used. The chemical composition of the  $\text{Cr}_3\text{C}_2/\text{Ni-Cr}$

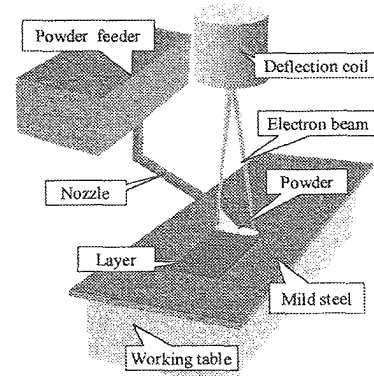


Fig. 1 Schematic drawing of the experimental apparatus

Table 1 Chemical composition of the  $\text{Cr}_3\text{C}_2/\text{Ni-Cr}$  alloy powder

$\text{Cr}_3\text{C}_2$	Ni-Cr	Particle size ( $\mu\text{m}$ )
50%	50%	8.8-55

alloy powder is shown in Table 1.

SS400 mild steel plate was used as the substrate. The powder was stably supplied at a constant feed rate of 0.4g/sec onto the substrate, which moved at a constant speed of 5mm/sec. The electron beam was oscillated at an amplitude of 20mm using a deflection coil and function generator. The  $\text{Cr}_3\text{C}_2/\text{Ni-Cr}$  alloy layer was formed on the substrate by the irradiation of the scanning electron beam at a high speed of 1600mm/sec.

The cladding layers were examined using optical microscope, electron probe micro analyzer (EPMA),

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	Beam current (mA)			
	25	30	35	40
Surface				
Cross-section				
HV	791	723	663	612

Fig. 2 Surface appearance, cross section and Vickers hardness of  $\text{Cr}_3\text{C}_2/\text{Ni-Cr}$  alloy cladding layers

Vickers hardness test (load: 300g, load time: 15sec), sand erosion test (ACT-JP method, Abrasive: mild steel, jet air pressure:  $5.0\text{kg/cm}^2$ ) immersion corrosion test (corrosion solution:  $\text{H}_2\text{SO}_4$  aqueous solution, corrosion time: 10 hours).

The surface appearance, cross section and Vickers hardness of  $\text{Cr}_3\text{C}_2/\text{Ni-Cr}$  alloy cladding layers formed at a variety of beam currents are shown Fig. 2. At a beam current of 25mA, porosity occurred and un-melted powder was recognized on the surface of the cladding layer. Layers without porosity were obtained at beam current of 30mA and 35mA. The layer's thickness was about  $100\mu\text{m}$ . At a beam current of 40mA, however, the substrate material mixed with the surface layer. The Vickers hardness of the  $\text{Cr}_3\text{C}_2/\text{Ni-Cr}$  alloy layer formed at a beam current of 25mA was 791HV. However, with increasing beam current, Vickers hardness of the  $\text{Cr}_3\text{C}_2/\text{Ni-Cr}$  alloy layer decreased. This is thought due to decomposition of the  $\text{Cr}_3\text{C}_2$  particles as a high-hardness component by electron beam irradiation.

Figure 3 shows the EPMA results with the  $\text{K}\alpha$  lines of Cr, Ni, Fe and C for the  $\text{Cr}_3\text{C}_2/\text{Ni-Cr}$  alloy layer formed at a beam current of 30mA. The  $\text{Cr}_3\text{C}_2$  particles are evenly dispersed within the matrix of the Ni-Cr alloy. There is no mixing of iron with the cladding layer.

Figure 4 shows the results of immersion corrosion tests and a sand erosion tests for SUS304 steel plate, SS400 steel plate and a  $\text{Cr}_3\text{C}_2/\text{Ni-Cr}$  alloy layer formed under the optimum beam condition of 35mA. The specimens were immersed in a 50%  $\text{H}_2\text{SO}_4$  aqueous solution for 10 hours and the amount of corrosion was measured. The corrosion rate of

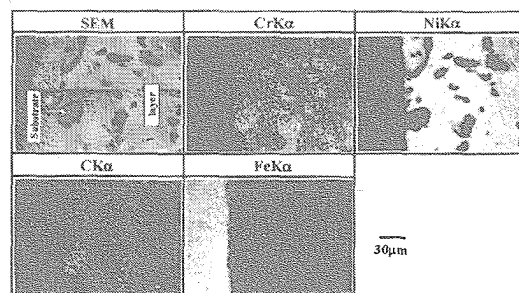


Fig. 3 EPMA results of  $\text{Cr}_3\text{C}_2/\text{Ni-Cr}$  alloy layer

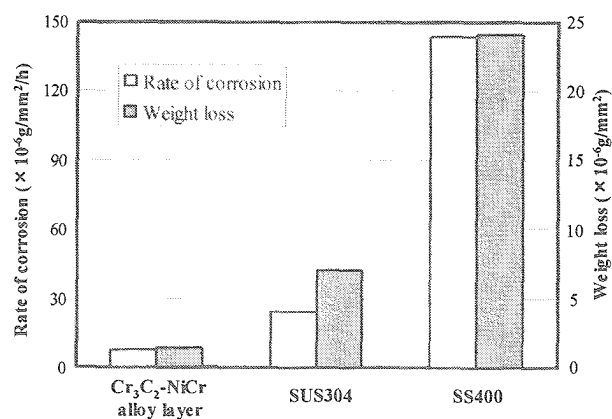


Fig. 4 The results of an immersion corrosion test and sand erosion test for  $\text{Cr}_3\text{C}_2/\text{Ni-Cr}$  alloy layer

the  $\text{Cr}_3\text{C}_2/\text{Ni-Cr}$  alloy layer was very low, being only 25% that of stainless steel and 7% that of mild steel. In the sand erosion test, the abrasive (mild steel, 450HV) was sprayed onto the specimen surface with a high speed, and the erosion resistance was determined by specimen weight loss.

The erosion loss of the  $\text{Cr}_3\text{C}_2/\text{Ni-Cr}$  alloy layer was also very low, being only 30% that of stainless steel and 5% that of mild steel.

By using the electron beam cladding method, a  $\text{Cr}_3\text{C}_2\text{-NiCr}$  alloy layer with a high hardness of 791HV as well as high erosion and corrosion resistance was successfully formed on mild steel plate.

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